

1. Explain or define the following terms: (15%)
  - (1). Heat (2). Cycle (3). Continuum (4). Adiabatic Process
  - (5). The Zeroth Law of Thermodynamics
  
2. Using the basic dimensions: Length(L), Time(T), Mass(M) and Temperature(K) to express the following dimension of physical quantities. (15%)
  - (1). Energy (2). Density (3). Pressure (4). Entropy (5). Power
  
3. A cyclic engine operates between two thermal energy reservoirs at 900°C and 21°C. The engine receives 1800 Btu of heat from high temperature reservoir. Calculate the maximum Work that can be obtained. (15%)
  
4. Prove the Clausius inequality. (20%)
  
5. Show that
  - (1).  $T ds = du + p dv$  (5%)
  - (2).  $T ds = dh - v dp$  (5%)

where T represents the temperature, s the specific entropy, u the specific internal energy, p the pressure, v the specific volume, h the specific enthalpy.
  
6. (1). Explain the meaning of the Joule-Thomson coefficient and derive the equation
 
$$m_J = \frac{1}{C_p} \left[ T \left( \frac{\partial n}{\partial T} \right)_p - n \right]$$
 by using the relationships  $h = h(T, p)$ ,  $T ds = dh - v dp$ ,
 
$$\text{and } \left( \frac{\partial n}{\partial T} \right)_p = - \left( \frac{\partial s}{\partial p} \right)_T. \quad (20\%)$$
  - (2). Show the  $m_J = 0$  for an ideal gas. (5%)